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Structures of Phenazostatins A and B, Neuronal Cell Protecting Substances of Microbial Origin

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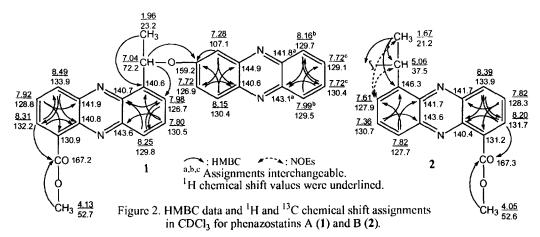
Abstract: Phenazostatins A (1) and B (2) were isolated from the culture broth of Streptomyces sp. 833 as neuronal cell protecting substances. These compounds were established as a member of the phenazine class of antibiotics on the basis of various spectroscopic analyses. Copyright © 1996 Elsevier Science Ltd

L-Glutamic acid, an excitatory amino acid, mediates neuronal degeneration in hippocampus through the generation of oxygen radicals after brain-ischemic attack ^{1,2}. It has been suggested that free radical scavengers may prevent neuronal cell death caused by L-glutamate³. In the course of our screening for substances to protect neuronal cells from the L-glutamate toxicity by using neuronal hybridoma N18-RE-105 cells, we isolated two phenazine compounds, phenazostatins A (1)⁴ and B (2), together with methyl saphenate^{5,6}. This paper describes the isolation and structural determination of 1 and 2 (Fig. 1).

A hexane extract of the broth filtrate (2 liters) of *Streptomyces* sp. 833 was purified by SiO₂ and Sephadex LH-20 column chromatographies followed by an ODS column chromatography cluted with 80% MeOH to give 1 (0.8 mg) and 2 (2.0 mg).

The molecular formula of 1 was established as $C_{28}H_{20}N_4O_3$ by HR-FAB mass spectroscopy (m/z 461.1626 (M+H)⁺ +1.2mmu) in combination with ¹H and ¹³C NMR data. The UV absorption maxima at 252 and 365 nm in McOH together with the presence of several nitrogen-conjugated ¹³C resonances between 140 and 145 ppm suggested that 1 was a typical phenazine compound. The IR absorption at 1730 cm⁻¹ and the ¹³C signal at 167.2 ppm revealed the presence of an ester group in 1. Two ABX proton spin systems (8.31, 7.92

Figure 1. Structures of phenazostatins A (1) and B (2).



and 8.49 ppm, and 7.98, 7.80 and 8.25 ppm) revealed that 1 had 1,6- and/or 1,9-disubstituted phenazine moieties. Since ¹H and ¹³C chemical shifts of 1 were in good agreement with those of methyl saphenate⁷, the structure of 1 was concluded to be a 1,6-disubstituted phenazine derivative. The structure of 1 was finally assigned by the HMBC data as shown in Fig. 2. It may be important to note that 1,6-dicarboxylic phenazines are ubiquitous among microbial metabolites⁶.

Compound 2 was previously reported by Umezawa *et al.*⁸ as an inhibitor of phosphodiesterase, but structural elucidation including stereochemistry remains still to be unequivocally solved. The molecular formula of 2 was determined to be C₃₂H₂₆N₄O₄ by HR-FAB mass spectroscopy (*m/z* 531.2040 (M+H)⁺ +0.6mmu). The UV and IR spectra of 2 were closely related to those of 1, implying that 2 was also a phenazine compound. The ¹H NMR spectrum revealed only 13 protons, while only 16 carbons were observed in the ¹³C NMR spectrum, indicating that 2 was a symmetric dimer. The ¹H and ¹³C chemical shifts of 2 are completely assigned by the HMBC data as shown in Fig. 2. The 1,6-disubstituted phenazine moiety for 2 was determined by comparing the ¹³C chemical shifts of 2 with those of methyl saphenate. Compounds 1 and 2 showing no optical rotation exist in nature as mixtures of enantiomers as suggested by Floss *et al.*⁶ and more than three stereoisomers⁹, respectively. The biological activities of 1 and 2 are now under investigation.

References

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- 4. Compound 1. Mp 220-225 °C; UV λ_{max} nm (ϵ) in McOH: 252 (86,000), 365 (18,000); IR (KBr): 1730, 1440, 1270, 1190, 1030, 760 cm⁻¹; [α]_D = 0°(c =0.035, CDCl₃); HRFAB-MS: m/z 461.1626 (MH⁺), $C_{28}H_{20}N_4O_3$ requires 461.1614. For compound **2**, see ref. 8.
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- 7. 13C chemical shifts of methyl saphenate in CDCl₃+CD₃OD (1:1): 168.1 (1-CO), 145.2 (6), 144.2 (9a), 142.2 (4a), 141.8 (5a), 141.1 (10a), 134.4 (4), 133.2 (2), 132.0 (8), 131.4 (1), 129.7 (3), 129.2 (9), 127.2 (7), 66.3 (12), 53.0 (11), 24.8 (13).
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- 9. NMR spectra of 2 showed the broad ¹H signals at 7, 8, 12 and 13 positions and two ¹³C peaks with low intensity for 7-CH. The HPLC analysis of 2 revealed three peaks, which were not resolved well.